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The performance of concrete exposed to marine environments: predictive modelling and use of laboratory/on site test methods


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ABSTRACT
This paper reports an approach by which laboratory based testing and numerical modelling can be combined to predict the long term performance of a range of concretes exposed to marine environments. Firstly, a critical review of the test methods for assessing the chloride penetration resistance of concrete is given. The repeatability of the different test results is also included. In addition to the test methods, a numerical simulation model is used to explore the test data further to obtain long-term chloride ingress trends. The combined use of testing and modelling is validated with the help of long-term chloride ingress data from a North Sea exposure site. In summary, the paper outlines a methodology for determining the long term performance of concrete in marine environments.

KEYWORDS: Chloride Ingress, Chloride Diffusivity, Electrical Resistivity, Concrete Testing, Chloride Modelling, Permit Ion Migration Test, Charge Passed, Performance Testing, Performance-based Specification

1. INTRODUCTION
Premature deterioration of reinforced concrete structures due to lack of durability and subsequent poor state of their health, has led to a significant part of the construction budget
being spent on repair and rehabilitation works. As a direct consequence, asset owners are often forced to take decisions to repair and maintain existing ailing infrastructure, as opposed to investing in new ones. Effective decision making in this regard requires systematic information about the state of health of an asset (or expected performance), an acceptable level of variance in this information and an effective maintenance strategy that is linked to its whole life value. In the case of concrete infrastructure, factors such as, for example, materials used, design and type of loading on the structure, its location and severity of the exposure condition, will all influence the decision making process as to the calculated state of health of the structure.

It is important, therefore, to specify the expected performance of a structure in addition to the guidelines given in standards, such as BS 8500, which detail the factors defined above. Some of the leading European and International research organisations such as RILEM Technical Committee 230-PSC, fib Task Group 8.10 and ACI Committee 365 are focusing on developing “Performance-based Specification for concrete”. The final technical report by Brite EuRam III DuraCrete [1] project provides an insight into the future of such specifications. Further articles on this topic can be found elsewhere [see, for example, 2-5]. As the quality of materials and exposure environments vary for different countries, it is also necessary to develop and refine the specifications so that they remain relevant to the practices of that country. The availability of test methods and understanding in the use of predictive models will also become critical in the general adoption of performance-based specification.

This paper details an approach by which laboratory and on site (in situ) testing can be further exploited with the help of a numerical model in order to determining the long term performance of concrete exposed to marine environments. The data presented in this paper are obtained from three research projects:

1) ChlorTest supported by European Union, Framework 5 [6];

2) PhD research by one of the authors (S.V. Nanukuttan) [7], and

3) Performance-based Testing Methodologies supported by Engineering Physical Science Research Council [8].

A range of test methods available for assessing the resistance to chloride ingress are reviewed and their repeatability compared. Numerical simulation of chloride movement into concrete is undertaken using a service life prediction model and the simulated data compared with on site
data obtained from a long-term marine exposure study. The main objective of this paper is to present and summarise developments in testing and modelling of concrete for chloride ingress and to illustrate how progress could be made in developing performance-based specifications.

2. MEASUREMENT OF RESISTANCE TO CHLORIDE INGRESS IN CONCRETE

Although the primary mechanism of chloride transport through unsaturated concrete cover is absorption, the accumulation of chlorides in this layer leads to further penetration of chlorides into the cover-zone by diffusion [9]. As a consequence, diffusion becomes the dominant mechanism of chloride transport at greater depths, which can be assessed in terms of the diffusion coefficient of the chloride ion. Different test methods are available to determine the diffusion coefficient, e.g. steady-state and non-steady-state chloride diffusion. As diffusion takes time to establish and tests are often tedious, migration tests are often used to quantify the movement of chlorides in concrete. The most common migration test uses an external electrical field to accelerate the flow of chloride ions, hence the coefficient determined using such a test is termed a migration coefficient. In addition, diffusivity is recognised as a term that represents chloride transport resistance of concrete, irrespective of the mechanism used to establish the flow of chlorides.

2.1 Experimental programme

Table 1 shows mix details of 9 typical concretes used across Europe. The concrete samples (large slabs) were cast as part of the EU funded ChlorTest programme[6]. Cores of 100mm diameter and 200mm length were extracted from the samples and these were transported to Queen’s University in a moist state and were kept at 20°C until the test date. Samples were wrapped with moist hessian to prevent drying. All data reported in Table 1 were generated by the tests carried out at Queen’s University. In addition to the mix design, the table also shows migration and electrical resistivity data from NT Build 443[10], NT Build 492[11] and bulk electrical resistivity [12] testing. Average chloride concentration at depths 5mm and 10mm from the surface from samples studied for NT Build 443[10] are also reported. Repeatability of the tests are discussed together with the long-term performance (readers interested in reproducibility of the tests may refer to ChlorTest[6] report where inter and intra laboratory comparisons were analysed).
Table 1 Details of concrete mixes (Quantities reported in kg/m³) and their chloride ingress resistance as measured by different test methods

2.2 Relationship between chloride penetration and concrete diffusivity assessed using different lab based test methods

Figs. 1 to 3 present the diffusivity of concrete (assessed using different laboratory-based tests) plotted against the quantity of chloride ions measured at 5 and 10 mm depths from the exposed surface. The chloride ion concentration at these depths was determined by the potentiometric titration method using powder samples collected from concrete specimen immersed in 2.8M NaCl solution for 35days. Data points represent nine different concrete mixes and the test age of concrete samples was >180days. Results presented in Figs. 1 to 3 indicate that the diffusivity assessed by the different test methods can be used with varying degree of accuracy to predict the chloride concentration at a particular depth.

![Graph](image)

**Fig. 1** - Concentration of chloride ions versus non-steady state diffusion coefficient (\(D_{nssd}\)) determined using NT Build 443[10] for 5mm and 10mm depths from the exposed face of concrete specimens.

Note: Age of testing is >180 days, Each data point represent a concrete mix refer to Table
Fig. 2 - Concentration of chloride ions versus non-steady state migration coefficient ($D_{\text{nsst}}$) for depths 5mm and 10mm from the exposed face of concrete specimens (obtained from Nordtest NT Build 492 [11]).

Note: Age of testing is >180 days, Each data point represent a concrete mix refer to Table 1 for mix details.

Figs. 2 and 3 suggest that useful information about the penetration of chloride ions can be obtained using rapid test methods. NT Build 492 [11] requires, on average, 24 hours to assess the diffusivity of concrete whereas electrical resistivity can be measured instantaneously. It is also worth noting that the electrical resistivity, in this case, was obtained from concrete specimens saturated with calcium hydroxide ($\text{Ca(OH)}_2$) solution. The procedure for carrying out the test can be found in references [12] and [6]. All these tests require concrete cores with a minimum thickness 50mm to be extracted from the structure which considerably limits the number of test that can be performed as frequent testing can leave the structure badly
disfigured. As a result, two on site methods were also included as part of this study which are outlined in the next section.

![Graph showing measured chloride concentration versus electrical resistivity of bulk concrete for 5mm and 10mm depths from the exposed face of concrete specimens. Age of testing is >180 days refer to Table 1 for details.](image)

**Fig. 3** - Concentration of chloride ions versus electrical resistivity of bulk concrete for 5mm and 10mm depths from the exposed face of concrete specimens. Age of testing is >180 days refer to Table 1 for details.

Note: Age of testing is >180 days, Each data point represent a concrete mix refer to Table 1 for mix details.

3. **ONSITE METHODS FOR ASSESSING CHLORIDE DIFFUSIVITY**

3.1 **Background of Permit Ion Migration Test (Permit)**

This is a unique, non-destructive, surface-based test developed on the principles of the steady state migration test. Initial developments of this test are detailed elsewhere [14] and further refinements during 2007-10 resulted in automation of the procedure and making it site/user
The test apparatus is shown in Fig. 4a. and a brief description of the apparatus and working principle are outlined below for the benefit of readers.

The Permit comprises two concentric cylinders placed on the concrete surface with the inner cylinder and outer cylinder containing 0.55M NaCl and de-ionised water, respectively. A potential difference is applied via a stainless steel cathode placed in the inner cylinder and mild steel anode placed in the outer cylinder. The test establishes a flow of chloride ions from inner cylinder through concrete to the outer cylinder. A conductivity sensor located in the outer cell monitors the conductivity of the electrolyte (initially de-ionised water) and this is converted to chloride concentration using relationship established by Castellote et al., [13]. The validity of conductivity/concentration relationship was established for the range of concretes using the Permit and is shown in Fig. 4b. The change in conductivity (therefore concentration) is used in the Nernst-Planck equation to obtain a migration coefficient ($D_{in\,situ}$) viz,

$$D_{in\,situ} = -\left(\frac{dc}{dt} \frac{T}{E} \frac{R}{zCF} \frac{L}{A}\right)$$  \hspace{1cm} \text{Eq. 1}

Where,

- $D_{in\,situ}$ = migration coefficient, m$^2$/s
- $\frac{dc}{dt}$ = rate of change of concentration in the anolyte, mol/m$^3$.s
- $T$ = absolute temperature (average during the steady state), K
- $R$ = universal gas constant (8.31 J/K.mol)
- $z$ = valency of the ions (-1 for chlorides)
- $C$ = concentration of ion source solution (0.55x10$^3$ mol/m$^3$)
- $F$ = Faraday’s Constant (9.65x10$^4$ c/mol)
- $E$ = electrical potential applied between the anode and the cathode (15, 30 or 60 V)
- $V$ = volume of the anolyte (6.5x10$^{-4}$ m$^3$)
L/A = ratio of the flow length to the exposed area (3.74 m⁻¹)

In addition to conductivity, the Permit also monitors the current (therefore charge passed) and temperature. A typical set of data from the Permit is presented in Figs. 4c and 4d. The reliability of the test was established by correlating the results with those obtained from other commonly used diffusion/migration test methods.

![Fig. 4a - The Permit Ion Permeability Tester (controller and the main body)](image)

![Fig. 4b - The relationship between conductivity and concentration in Permit ion migration test (all mixes). Refer to Table 2 for mix details.](image)

\[ y = 6.01x \]
\[ R^2 = 0.94 \]
3.2 $D_{\text{in situ}}$ and other measures of chloride transport

The objectives of this study were to establish:
(1) the validity of $D_{in situ}$ in assessing diffusivity for a range of concretes;

(2) the effect of supplementary cementitious materials on $D_{in situ}$; and,

(3) the usefulness of current and charge passed as a measure of chloride diffusivity.

Eight concrete mixes containing Portland cement (CEM I), pulverised fuel ash (pfa), ground granulated blast-furnace slag (ggbs) and microsilica (ms) were used. Details of the mix design, test age are presented in Table 2.

Table 2. Details of mixes used in this experimental programme

3.2.1 Relationship between $D_{in situ}$ and steady state migration coefficient

A typical steady state migration test was carried out in this investigation that resembles the test proposed by Castellote [13]. The only deviation was that the concentration of catholyte (NaCl) solution was 0.55M instead of 1.0M. The relationship between $D_{in situ}$ and the coefficient determined using the steady state migration test is presented in Fig. 5. There is a high degree of correlation between the two migration tests ($R^2=0.93$), showing that the chloride diffusivity determined with the Permit ion migration test is as accurate as the more established steady state migration test. Basheer et al., [14] reported similar result for CEM I cement mixes and the results in Fig. 5 are for mixes containing a range of different cementitious materials. It can be concluded, therefore, that the Permit ion migration test can be used to determine the chloride diffusivity irrespective of the binder type in concretes. Moreover, the $D_{in situ}$ values were determined within 1 day compared to 7 days required for determining the steady state migration coefficient.
Fig. 5 – Graph demonstrating the usefulness of $D_{in \text{ situ}}$ in determining the steady state migration coefficient. (Legend shows, mix reference as in Table 2 + age of concrete at test, Example: 0.45 opc 28D – 0.45 w/b opc as in Table 2 tested at 28 days).

3.2.2 Relationship between $D_{in \text{ situ}}$ and non-steady state migration coefficient

NT Build 492 test is commonly used by the industry to quantify diffusivity, therefore $D_{nssm}$ obtained from this test was compared with $D_{in \text{ situ}}$. Fig. 6 shows that these two parameters have a linear relationship ($R^2 = 0.85$). Several factors are known to influence the relationship between a steady state and a non-steady state test result. This can be explained with the equation suggested by Nilsson et. al., [9]:

$$D_{\text{non-steadystate}} = \frac{D_{\text{steadystate}}}{\varepsilon \left(1 + \frac{\partial \bar{C}_b}{\partial \bar{C}_f}\right)}$$  \hspace{1cm} \text{Eq. 2}

Where $\varepsilon$ is the porosity and $\frac{\partial \bar{C}_b}{\partial \bar{C}_f}$ is the binding capacity of concrete.

This would imply that the relationship will be affected by both binding capacity and porosity and hence a larger scatter will be expected if experiment is repeated with different binder contents, type of binder or water-binder ratio.
Fig. 6 - Graph demonstrating the usefulness of $D_{in\,situ}$ in determining the non-steady state migration coefficient. (Legend shows, mix reference as in Table 2 + age of concrete at test, Example: 0.45 opc 28D – 0.45 w/b opc as in Table 2 tested at 28 days).

3.2.3 Effect of type of binder and w/b on $D_{in\,situ}$

The results indicate that $D_{in\,situ}$ is sensitive to both binder type and w/b as shown in Fig. 7. The use of supplementary cementitious materials generally decreased the $D_{in\,situ}$ at both w/b ratios investigated. Whereas ggbs and ms mixes exhibit the lowest $D_{in\,situ}$, the pfa mixes had the highest $D_{in\,situ}$. This could be due to the slow initial hydration of pfa particles. The trend shown in Fig. 7 is similar to that observed by Basheer et al., [14]. Therefore, it can be stated that the $D_{in\,situ}$ was able to distinguish the variations in the mixes and in a manner similar to that by most migration tests.
3.3 Peak current, charge passed and chloride diffusivity

In addition to estimating conductivity (concentration), the Permit also monitors the current and temperature evolution. Current evolution follows an unusual behavior as shown in Fig. 4c. In order to further explore this behaviour, the peak value of the current was compared to that of $D_{\text{in situ}}$.

3.3.1 Relationship between $D_{\text{in situ}}$ and the peak current—Fig. 8 shows that the relationship between $D_{\text{in situ}}$ and the peak current is linear with a coefficient of regression of $R^2=0.82$. As shown in Fig. 4c, the peak value of the current is obtained when the flow of chloride ions reach steady state. Therefore it should be is possible to determine the diffusivity with the help of transference number and Nernst-Einstein equation as explained by Andrade [15]. This is outside the scope of this paper and will not be discussed.

3.3.2 Relationship between $D_{\text{in situ}}$ and the charge passed—There is no apparent relationship between $D_{\text{in situ}}$ and charge passed for the first 6 hours as seen in Fig. 9. This is to be expected as all ions present in the surface concrete will be taking part in the charge transfer process therefore attributing this to just chloride ions is not correct. The relationship improves marginally when the charge passed until steady state is compared to that of $D_{\text{in situ}}$ (refer to Fig. 10).
As the flow reaches steady state in a migration test, the current approaches its peak value, therefore the charge passed during the steady state will have a better relationship with chloride flow. To study this further, the rate of change of charge passed (average current) during steady state was plotted against the rate of change of chloride concentration, the result is presented in Fig. 11.

The regression relationship in Fig. 11 is represented by Eq. 3, with a regression coefficient (R²) of 0.85.

\[
\frac{dQ}{dt} = 210.81 \frac{dc}{dt}
\]

(Eq. 3)

where

\[
\frac{dQ}{dt}
\]

is the rate of change of charge during the steady state, coulombs/s,

\[
\frac{dc}{dt}
\]

is the rate of change of chloride concentration during the steady state, mol/m³.s,

Equation 3 suggests that the rate of change of chloride concentration can be estimated with 85% confidence using the value of the rate of change of charge. A similar trend was reported by Yang [16], where the author found that the charge passed during steady state flow was more comparable to the chloride penetration resistance of concrete than the first 6 hours charge passed. The implication of this is that by measuring the current, the rate of change of chloride concentration can be estimated in the Permit. Such a change would not only simplify the test procedure, but also reduce the cost of manufacturing the test equipment.
Fig. 8 - Relationship between $D_{\text{in situ}}$ and the peak current in Permit ion migration test.

\[ y = 2.14x - 1124 \]
\[ R^2 = 2 \times 10^{-0.05} \]

Fig. 9 - Relationship between $D_{\text{in situ}}$ and the charge passed during the initial 6 hours in Permit ion migration test. The graph only reports data from 28 day and 56 day tests.

\[ y = 219.51x + 1047 \]
\[ R^2 = 0.35 \]

Fig. 10 - Relationship between $D_{\text{in situ}}$ and the charge passed until steady state in Permit ion migration test. The graph only reports data from 28 day and 56 day tests.
3.4 Wenner four probe resistivity and chloride diffusivity

Published literature suggests that resistivity (bulk resistivity) of either water saturated or ionically (NaCl or Ca(OH)₂) saturated concrete specimens can be used to predict the steady or non-steady state migration coefficients [12, 17]. In this section the suitability of a surface based electrical resistivity test, i.e., Wenner four probe test is studied for determining the chloride diffusivity. The relationship between steady state migration coefficient and inverse of the Wenner four probe surface resistivity $1000/\rho_{Wenner}$ is presented in Fig. 12, which shows a linear relationship between these two test parameters. The relationship is represented by the following equation, with an $R^2$ value of 0.7:

$$D_{ssm} = \frac{260}{\rho_{Wenner}} \quad (Eq. 4)$$

Where

$\rho_{Wenner}$ is in ohm.m,

and $D_{ssm}$ is in m²/s (without the multiplier $10^{-12}$).
The relationship is likely to be affected by the level of contact made by the wooden plugs with concrete surface in Wenner four probe resistivity test. Further, any small defects at the point of contact of the concrete surface and degree of saturation could also affect the resistivity value. If bulk conductivity (inverse of bulk resistivity) is assessed instead of Wenner test, the relationship is highly likely to improve as the resistivity is measured across the whole of the test specimen [6, 12, 17]. In any case, the resistivity will be influenced by the nature of the pore solution [18], therefore saturating the test surface will help to get consistent results.

The relationship between the $D_{\text{in situ}}$ and the inverse of the Wenner probe resistivity, $1000/\rho_{\text{Wenner}}$ is presented in Fig. 13. The regression relationship is presented by the following equation:

$$D_{\text{in situ}} = \frac{140}{\rho_{\text{Wenner}}}$$  \hspace{1cm} (Eq. 5)

where,

$D_{\text{in situ}}$ is in m$^2$/s (without the multiplier $10^{-12}$),

$\rho_{\text{Wenner}}$ is in ohm.m.

The coefficient of determination $R^2$ obtained in this case was 0.72, higher value than that obtained for the relationship in Eq. 4. The limitations discussed in previous section will also affect the relationship with $D_{\text{in situ}}$. The improvement is due to the fact that both tests were performed on same concrete samples, i.e., water saturated. With this relationship, end-users will be able to rapidly test the surface of concrete using Wenner test and relate these results that to either a $D_{\text{in situ}}$ or $D_{\text{ssm}}$ performed on the same concrete. It is necessary, however, to establish this relationship for the concrete being tested as several factors including the pore solution composition, curing and type of aggregate will also affect the relationship.
Fig. 12 - Relationship between Wenner four probe electrical resistivity and steady state migration test for a range of concrete samples. Both steady state migration and Wenner test carried out on concrete saturated with water prior to testing. Wenner test is performed on 230x230x100mm concrete slabs used for Permit.
3.5 Summary of on site tests for assessing diffusivity

It was shown that both tests discussed above can quantify diffusivity through their respective parameters. The more rapid $\rho_{\text{Wenner}}$ is affected by the condition of the test surface and therefore the scatter is large. The scatter expected on site will be greater than that reported in Figs. 12 and 13 due to the uncontrolled nature of test surface. For concretes contaminated with ions, or with different pore solution composition, it would be best to use the Permit or a steady state migration test rather than the electrical resistivity based test.

Results presented in Figs. 5-11 show that the Permit can be used as an alternative to conventional and more commonly used laboratory-based test methods. The following advantages have been identified for the Permit:

a) The test can be performed in situ and there is no need to remove cores for testing.

b) The test procedure is automated therefore it reduces,
   i. human error and does not require a skilled operator to perform the test,
   ii. the overall cost of carrying out the test or performance assessment.

c) The test can be performed in combination with the more rapid electrical resistivity test (Figs. 12 and 13) which allows an overall appraisal of the structure within a short period of time, thereby reducing the cost of carrying out condition assessment. This is dealt in detail in the next section.

d) The charge passed obtained by the test during the first 6 hours show no apparent correlation. However, the charge passed during the steady state appears to correlate well with the $D_{\text{in situ}}$. This is in agreement with results reported elsewhere. Such simplification could provide an alternative way for conducting the test using a more simplified version of the Permit and also help to modify the ASTM C1202 [19] test to yield more reliable coefficient that can represent chloride diffusivity.

4. EFFECT OF CONCRETE MIX PROPERTIES ON LONG TERM PERFORMANCE
Several test methods which can assess the resistance of concrete to chloride ingress were identified above. It is important to understand the repeatability and scope of the results in order for the test to be used for qualifying concretes. The results in Table 1 identify the beneficial effects of using supplementary materials, such as pfa, ggbs and ms, and the influence of w/b on chloride ingress. Most of the results are, on average, ±20% from the median, hence it can be concluded that the tests are repeatable with 20% variability. To study the scope of these results it is necessary either to study the long-term behaviour of these concrete mixes in a field exposure environment or to simulate the behaviour in a given environment. The former would require long-term study with considerable investment and resources, whereas the latter would depend heavily on the accuracy of the numerical model used for predicting the behaviour.

The approach used in this paper is to consider both the aspects. The long-term performance data from a structure exposed to a marine environment (North Sea) is used to validate the numerical model used for prediction. The second aspect is to use the test results in Table 1 along with the validated numerical model to predict the behaviour of different concrete mixes in the same environment.

4.1 Long-term performance study on concrete specimens exposed to North Sea

Data from a long-term study conducted on three ordinary Portland cement concrete pier stems exposed to tidal, splash and atmospheric conditions in North Sea are presented below. The concrete mix details are reported in Table 3. Chloride concentrations from various depths (termed as chloride profile) were determined several times for a period up to 7 years and then after 18 years. General location of the piers and annual temperature variation at the site is as shown in Figs. 14 and 15 respectively. Fig. 16 presents the chloride profiles determined after 1.17, 3.17, 6.17 and 18 years exposure and located below mid-tide level - XS2 environment according to [20] (i.e. remaining saturated). Further details on these pier-stems, surface chloride concentrations and chloride profiles are available elsewhere [21], therefore only data relevant to the modelling are detailed here.

**Table 3 Mix details and test results for Dornoch pier stems**
Several service life prediction models were considered as part of the wider study. However, only data from numerical simulations made using ClinConc service life model [22] are reported in this paper as, this model was selected based on the recommendations by the ChlorTest [6] project, which reviewed different test methods and service life models.

The measured and numerically simulated chloride profiles are presented in Figs. 17-19. The upper and lower lines indicate the level of pessimistic and optimistic variations due to the positive and negative disparity, respectively, in input parameters including Dnssm (+/- 2\times 23\% as the reproducibility variation [6] for a confidence level of 95%), the chloride concentration
in sea water (assuming +/- 10% as the annual mean variation) and the exposure temperature (assuming +/- 1 °C as the annual mean variation). Figs. 17 and 18 show that the numerical simulations can predict the chloride profile with a high degree of accuracy. At the age of 18 years (Fig. 19), the simulation underestimates the chloride ion content at depths greater than 30mm; however, the data points are within the upper limit considered. It can be seen from Figs. 17-19 that the chloride ingress depth is strongly affected by the variations of the input parameters. A more precise service life design can be achieved by increasing the confidence in the input parameters. For example, by carrying out tests to determine $D_{nsm}$ on concrete cured in same condition as the real structure.

![Fig. 17 - The real and predicted chloride concentration after 1.17 years of exposure.](image1)

![Fig. 18 - The real and predicted chloride concentration after 7.17 years of exposure.](image2)

![Fig. 19 - The real and predicted chloride concentration after 18 years of exposure.](image3)
points indicate real data collected from the North Sea exposure site]

Table 4. List of variables used in the ClinConc Model

4.2 Guidelines for selecting concrete mixes for marine exposures

Based on the non-steady state migration coefficient ($D_{nssm}$) in Table 1 and ClinConc model that takes $D_{nssm}$ as one of the input parameter, chloride profiles were simulated for the different concrete mixes exposed to the North Sea environment. Fig. 20 shows the chloride profiles after 50 years of exposure to tidal low-level (XS2 environment) exposure zone in North Sea. This will help to identify the suitability of mixes specified as per BS 8500 [20]. For XS2 environment, BS8500 allows w/b of $\leq 0.55$, most cement combinations and a minimum binder content of $320\text{kg/m}^3$. Fig. 20 shows that even $0.5\text{w/b}$ concrete studied here performs poorly in such environment. It is necessary, therefore, to undertake similar studies in order to identify the performance of concrete for various exposure environments and adoption of a performance-based approach in specifying concrete. Test results such as $D_{nssm}$ identified in Table 1 can be used in defining performance-based specifications for concretes. As an example, in order to keep the chloride concentration at the level of reinforcement that is at a depth of 50mm from the exposure surface to a value below 0.5 % by mass of binder, one could use the ggbs 0.42 concrete or, indeed, any concrete which has a diffusivity $D_{nssm} < 1 \times 10^{-12} \text{ m}^2\text{s}$ at an age of 180days. The above guideline is conservative and only considers the time to corrosion initiation; the actual service life will depend on the corrosion propagation phase and the acceptable level of deterioration.
CONCLUDING REMARKS

The applicability, scope and repeatability of various laboratory based test methods for assessing the chloride penetration resistance have been demonstrated. The validity of the Permit and Wenner resistivity test as on site methods for quantifying chloride transport in concrete was also established. It was identified that the use of the Permit eliminates the need for extracting concrete cores for the laboratory determination of the chloride diffusivity.

It was shown that the data from chloride testing can be used to predict the chloride concentration in a given exposure environment and at a given time through the use of a numerical simulation model. The accuracy of the prediction was also verified by comparing the predicted data against field data obtained from a long-term study. It was found that over the initial 7-years exposure the predicted and measured values were in good agreement; however, there was an underestimation of chloride content at depths > 30mm at 18 years. This will require further refinement of the model.
The combined use of testing and modelling for determining the long term performance of concretes was demonstrated and it should now be possible to adopt this approach to develop performance-based specification for concrete exposed to marine environments. Further challenges include, improving the accuracy of the model for later years, simplifying the modelling process and estimating the chloride threshold concentration that initiate corrosion.

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REFERENCES


Table 1 Details of concrete mixes (Quantities reported in kg/m$^3$) and their chloride ingress resistance as measured by different test methods

<table>
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<th>Mix designation</th>
<th>opc 0.35</th>
<th>opc 0.45</th>
<th>opc 0.50</th>
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<th>ms 0.42</th>
<th>pfa 0.42</th>
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<td>CEM I-42.5 N</td>
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<td>(18% PFA) CEM II/A-V 42.5 R</td>
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<td></td>
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<td>410</td>
<td>340</td>
<td>410</td>
<td>350</td>
</tr>
<tr>
<td>Microsilica</td>
<td></td>
<td></td>
<td></td>
<td>21</td>
<td>20.5</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Water</td>
<td>157.5</td>
<td>180</td>
<td>200</td>
<td>168</td>
<td>172.2</td>
<td>172.2</td>
<td>153</td>
<td>172.2</td>
<td>157.5</td>
</tr>
<tr>
<td>Fine Aggregate (Min size 75μm)</td>
<td>904 (≤8mm)</td>
<td>742 (≤6mm)</td>
<td>920 (≤8mm)</td>
<td>842.5 (≤8mm)</td>
<td>897 (≤8mm)</td>
<td>901 (≤8mm)</td>
<td>62 (≤2mm)</td>
<td>603 (2-4mm)</td>
<td>901 (≤8mm)</td>
</tr>
<tr>
<td>Coarse Aggregate</td>
<td>904 (5-10mm)</td>
<td>1030 (5-16mm)</td>
<td>816 (5-10mm)</td>
<td>842.5 (8-16mm)</td>
<td>897 (10-15mm)</td>
<td>901 (10-15mm)</td>
<td>619 (4-12mm)</td>
<td>555 (12-25mm)</td>
<td>901 (5-10mm)</td>
</tr>
<tr>
<td>Superplasticiser % of cement</td>
<td>CemFlux Bro 1.0</td>
<td>Melcret 222 4.8</td>
<td>Cementa 92M 3.4</td>
<td>CemFlux Bro 0.5</td>
<td>CemFlux Bro 0.5</td>
<td>Rheobuild 1000 4.1</td>
<td>CemFlux Bro 0.5</td>
<td>Cretoplast 3.9</td>
<td></td>
</tr>
<tr>
<td>water/binder (w/b)</td>
<td>0.35</td>
<td>0.45</td>
<td>0.5</td>
<td>0.4</td>
<td>0.42</td>
<td>0.42</td>
<td>0.45</td>
<td>0.42</td>
<td>0.45</td>
</tr>
<tr>
<td>Age at test (years)</td>
<td>~0.5</td>
<td>~1.0</td>
<td>~0.5</td>
<td>&gt;1.0</td>
<td>~0.5</td>
<td>~0.5</td>
<td>&lt;1.0</td>
<td>~0.5</td>
<td>&lt;1.0</td>
</tr>
</tbody>
</table>

Measurable performance indicators (chloride diffusivity/bulk electrical resistivity results)
### Diffusivity/Resistivity

<table>
<thead>
<tr>
<th></th>
<th>opc 0.35</th>
<th>opc 0.45</th>
<th>opc 0.50</th>
<th>ms 0.40</th>
<th>ms 0.42</th>
<th>pfa 0.42</th>
<th>pfa 0.45</th>
<th>ggbs 0.42</th>
<th>ggbs 0.45</th>
</tr>
</thead>
<tbody>
<tr>
<td>$D_{nssd} \times 10^{-12}$ m$^2$/s (standard error)</td>
<td>5.11 (±0.56)</td>
<td>14.63 (±3.74)</td>
<td>16.56 (±1.82)</td>
<td>1.61 (±0.62)</td>
<td>4.88 (±0.58)</td>
<td>1.44 (±0.27)</td>
<td>7.38 (±2.43)</td>
<td>1.31 (±0.16)</td>
<td>3.19 (±1.35)</td>
</tr>
<tr>
<td>Chloride concentration (% by mass of binder)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>at 5mm from surface</td>
<td>0.27</td>
<td>0.29</td>
<td>0.44</td>
<td>0.16</td>
<td>0.32</td>
<td>0.11</td>
<td>0.42</td>
<td>0.08</td>
<td>0.14</td>
</tr>
<tr>
<td>at 10mm from surface</td>
<td>0.04</td>
<td>0.17</td>
<td>0.24</td>
<td>-</td>
<td>0.05</td>
<td>-</td>
<td>0.18</td>
<td>0</td>
<td>0.03</td>
</tr>
<tr>
<td>$D_{nssm} \times 10^{-12}$ m$^2$/s (standard error)</td>
<td>6.00 (±1.24)</td>
<td>15.00 (±3.02)</td>
<td>16.70 (±0.99)</td>
<td>1.90 (±0.07)</td>
<td>6.90 (±0.50)</td>
<td>1.70 (±0.13)</td>
<td>3.70 (±0.54)</td>
<td>1.00 (±0.05)</td>
<td>2.20 (±0.25)</td>
</tr>
<tr>
<td>$\rho_{bulk}$ (ohm.m) (standard error)</td>
<td>176 (±16)</td>
<td>187 (±22)</td>
<td>56 (±13)</td>
<td>427 (±21)</td>
<td>236 (±31)</td>
<td>324 (±40)</td>
<td>291 (±5)</td>
<td>838 (±160)</td>
<td>470 (±17)</td>
</tr>
</tbody>
</table>

CemFlux Bro is polycarboxylether based superplasticiser; Melcret 222 and Rheobuild 1000 are both naphthalene based superplasticisers; Cretoplast is a water reducing superplasticiser; Cementa 92M is melamine formaldehyde based superplasticiser; $D_{nssd}$ is the coefficient from non-steady diffusion test – NT Build 443; $D_{nssm}$ is the coefficient from non-steady migration test – NT Build 492; $\rho_{bulk}$ is the saturated bulk electrical resistivity – Andrade et al., (2000)
Table 2. Details of mixes used in this experimental programme

<table>
<thead>
<tr>
<th>Mix reference</th>
<th>w/b</th>
<th>Water</th>
<th>Portland cement</th>
<th>Cement replacements</th>
<th>Course Aggregate</th>
<th>10mm</th>
<th>6mm</th>
<th>sand</th>
<th>Superplasticiser (% by mass of cement)</th>
<th>28-day Compressive strength (N/mm²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.45 opc</td>
<td>0.45</td>
<td>180.0</td>
<td>400.0</td>
<td>0</td>
<td>1030.0</td>
<td>742.0</td>
<td></td>
<td></td>
<td>28</td>
<td>34.5</td>
</tr>
<tr>
<td>0.45 pfa</td>
<td>0.45</td>
<td>153.0</td>
<td>207.4</td>
<td>132.6 (pfa)</td>
<td>1174.0</td>
<td>665.0</td>
<td></td>
<td></td>
<td>28</td>
<td>3.4 28.0</td>
</tr>
<tr>
<td>0.40 ms</td>
<td>0.40</td>
<td>168.0</td>
<td>399.0</td>
<td>21.0 (ms)</td>
<td>842.5</td>
<td>842.5</td>
<td></td>
<td></td>
<td>28</td>
<td>3.4 35.5</td>
</tr>
<tr>
<td>0.45 ggbbs</td>
<td>0.45</td>
<td>157.5</td>
<td>84.0</td>
<td>266.0 (ggbbs)</td>
<td>1040.0</td>
<td>860.0</td>
<td></td>
<td></td>
<td>28</td>
<td>4.5 39.6</td>
</tr>
<tr>
<td>0.52 opc</td>
<td>0.52</td>
<td>208.0</td>
<td>399.9</td>
<td>0</td>
<td>1199.7</td>
<td>659.9</td>
<td></td>
<td></td>
<td>28</td>
<td>0.9 21.0</td>
</tr>
<tr>
<td>0.52 pfa</td>
<td>0.52</td>
<td>204.0</td>
<td>274.6</td>
<td>117.7 (pfa)</td>
<td>1176.7</td>
<td>647.2</td>
<td></td>
<td></td>
<td>28</td>
<td>0.9 13.0</td>
</tr>
<tr>
<td>0.52 ms</td>
<td>0.52</td>
<td>206.8</td>
<td>357.9</td>
<td>39.8 (ms)</td>
<td>1193.0</td>
<td>656.1</td>
<td></td>
<td></td>
<td>28</td>
<td>0.9 37.0</td>
</tr>
<tr>
<td>0.52 ggbbs</td>
<td>0.52</td>
<td>206.7</td>
<td>198.8</td>
<td>198.8 (ggbbs)</td>
<td>1192.7</td>
<td>656.0</td>
<td></td>
<td></td>
<td>28</td>
<td>0.9 21.0</td>
</tr>
</tbody>
</table>

Note:  
- w/b - water to binder ratio,  
- ms - micro silica,  
- pfa - pulverised fuel ash,  
- ggbbs - ground granulated blast furnace slag,  
- SP - superplasticiser – Chemcrete S725; mixes with w/b ≤0.45 was tested at 28days (and 2 years), mixes with w/b 0.52 was tested at 56days.

Table 3. Mix details and test results for Dornoch pier stems

<table>
<thead>
<tr>
<th>Mix</th>
<th>Cement kg/m³</th>
<th>20mm kg/m³</th>
<th>10mm kg/m³</th>
<th>Fines kg/m³</th>
<th>Plasticiser L/m³</th>
<th>w/c</th>
<th>F28 MPa</th>
</tr>
</thead>
<tbody>
<tr>
<td>Plain</td>
<td>460</td>
<td>700</td>
<td>350</td>
<td>700</td>
<td>3.60*</td>
<td>0.4</td>
<td>66</td>
</tr>
</tbody>
</table>

* Sika FR
### Table 4. List of variables used in the ClinConc Model

<table>
<thead>
<tr>
<th>$D_{\text{ssm}}$ m$^2$/s</th>
<th>Chloride concentration of sea water</th>
<th>Air Content</th>
<th>Na$_2$O equivalent</th>
<th>Initial chloride concentration</th>
<th>Curing and exposure temperature</th>
<th>Activation Energy</th>
</tr>
</thead>
<tbody>
<tr>
<td>$9(\pm 4.1) \times 10^{-12}$ (estimated using Table 1)</td>
<td>19.6g/L</td>
<td>3% by volume</td>
<td>0.41% Binder</td>
<td>0.05% by mass of binder</td>
<td>Curing 15°C for first 14 days, Exposure – 10°C annual average</td>
<td>36000 J/mol</td>
</tr>
</tbody>
</table>